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Supporting Information

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Noduliprevenone, a Novel Heterodimeric Chromanone with Cancer Chemopreventive Potential

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Table S1: NMR spectral data for compound **1**.

no. ^a	$\delta^{13}\text{C}$ (mult.) ^b	$\delta^1\text{H}$ (J in Hz) ^b	COSY ^{b,d}	HMBC ^{b,e}	NOESY ^{b,d}	sel. NOE ^{c,d}
1'	161.7 (C)					
2'	111.0 (CH)	6.49 (s)	11'	1', 4', 4a', 9', 9a', 11'	OH-1', 11'	OH-1', 11'
3'	151.6 (C)					
4'	115.0 (C)					
4a'	157.8 (C)					
5'	81.4 (CH)	4.89 (br t, 5.5)	6'	4a', 6', 7', 8', 8a', 10a'	6', 8a'α	6', 8a'α
6'	22.4 (CH ₂)	2.37 (m)	5', 7'α, 7'β	5', 7', 8', 10a'	5'	5', 7'α, 7'β, 8a'β
7'	26.9 (CH ₂)	2.13 (Hβ m)	6', 7'α	5', 6', 8'		6', 7'α
		1.85 (Hα t, 9.9)	6', 7'β	5', 6', 8'	OH-1	OH-1, 7'β
8'	176.0 (C)					
8a'	40.0 (CH ₂)	3.37 (Hβ d, 17.2)	8a'α	5', 9', 9a', 10a', 12'		8a'α
		2.97 (Hα d, 17.2)	8a'β	5', 9', 9a', 10a', 12'	5'	5', 8a'β
9'	195.4 (C)					
9a'	106.7 (C)					
10a'	86.2 (C)					
11'	20.8 (CH ₃)	1.96 (s)	2'	2', 3', 4', 4a', 9a'	2', OH-1	2', 11
12'	169.8 (C)					
13'	53.5 (CH ₃)	3.70 (s)		12'	11	11, 8a'α

OH-1'		11.47 (s)		1', 2', 3', 9', 9a'	2'	2'
1	159.5 (C)					
2	116.6 (C)					
3	151.6 (C)					
4	109.8 (CH)	6.43 (s)	11	1, 2, 9, 9a, 11	11	11
4a	159.8 (C)					
5	73.7 (CH)	4.14 (br d, 10.6)	OH-5, 6	6, 7, 8a, 10a	6, 7, 8a α , 8a β	6, 7, 8a α , 13
6	26.7 (CH ₂)	1.73 (m)	5, 7	5, 7, 8	5	5
7	30.9 (CH ₂)	2.54 (m)	6	5, 6, 8	5	5
8	173.8 (C)					
8a	37.9 (CH ₂)	3.33 (H α d, 17.2) 3.04 (H β d, 17.2)	8a β 8a α	5, 9, 9a, 10a, 12 5, 9, 9a, 10a, 12	5 5, 6	8a β 6, 8a α
9	197.7 (C)					
9a	105.9 (C)					
10a	88.1 (C)					
11	21.0 (CH ₃)	2.12 (s)	4	2, 3, 4, 4a, 9a	4, 13'	4, 11', 13'
12	170.9 (C)					
13	53.5 (CH ₃)	3.77 (s)		12		
14	51.6 (CH ₃)	3.63 (s)		8		
OH-1		11.94 (s)		1, 2, 3, 9, 9a	7'a, 11'	
OH-5		5.05 (br d, 4.8)	5		5	

^aPosition of carbon atom. ^b[D₆]Acetone, 300/75.5 MHz. ^c[D₆]Acetone, 500/75.5 MHz. ^dNumbers refer to proton resonances. ^eNumbers refer to carbon resonances.

Supporting information – General Experimental

All NMR spectra were recorded on Bruker Avance 500 DRX or 300 DPX spectrometers in [D₆]acetone or CDCl₃. Spectra were referenced to residual solvent signals. UV and IR spectra were obtained employing Perkin-Elmer Lambda 40 and Perkin-Elmer Spectrum BX instruments, respectively. HRESIMS were recorded on Bruker Daltonics' micrOTOF-Q instrument. Optical rotation was measured on a Jasco DIP 140 polarimeter.

Circular Dichroism Spectroscopic Measurements. CD spectra of compound **1** and **2** were recorded at room temperature on a Jasco J-810 spectrophotometer. Samples were measured in 1 cm-cuvettes at concentration of 0.3 mmol L⁻¹.

Modified Mosher Derivatization

Preparation of the acid chlorides of (R)- and (S)-MPA. Oxalyl chloride (103.7 μ L, 1.2 mmol) was added to a solution of the corresponding MPA (20 mg, 0.12 mmol) and DMF (0.94 μ L, 0.012 mmol) in hexane at room temperature. After one day, the solvent was vacuum-baked to give 100% of the product MPA-Cl (22.2 mg, 0.12 mmol).

Preparation of the (R)- and (S)- MPA esters. The corresponding MPA-Cl (4.27 mg, 23.15 μ mol) was dissolved in 5 mL of CH₂Cl₂ and added to a solution of compound **1** (3.1 mg/ 4.63 μ mol for *R*-derivatization; 2.4 mg/ 3.6 μ mol for *S*-derivatization), Et₃N (7.70 μ L, 55.56 μ mol) and DMAP (0.57 mg, 4.63 μ mol) as catalyst. After 20 min reaction time the obtained products were evaporated under vacuum and further purified by HPLC using a Merck-Hitachi system consisting of a L-6200 A pump, a L-4500 A photodiode array detector and a D-6000 A interface. The separation was performed with a RP18 column (Macherey-Nagel Nucleodur Sphinx RP, 5 μ m, 250 x 4.6 mm) and a mobile phase (1.2 mL/min) consisting of MeCN/H₂O 64/36. The main peaks in the

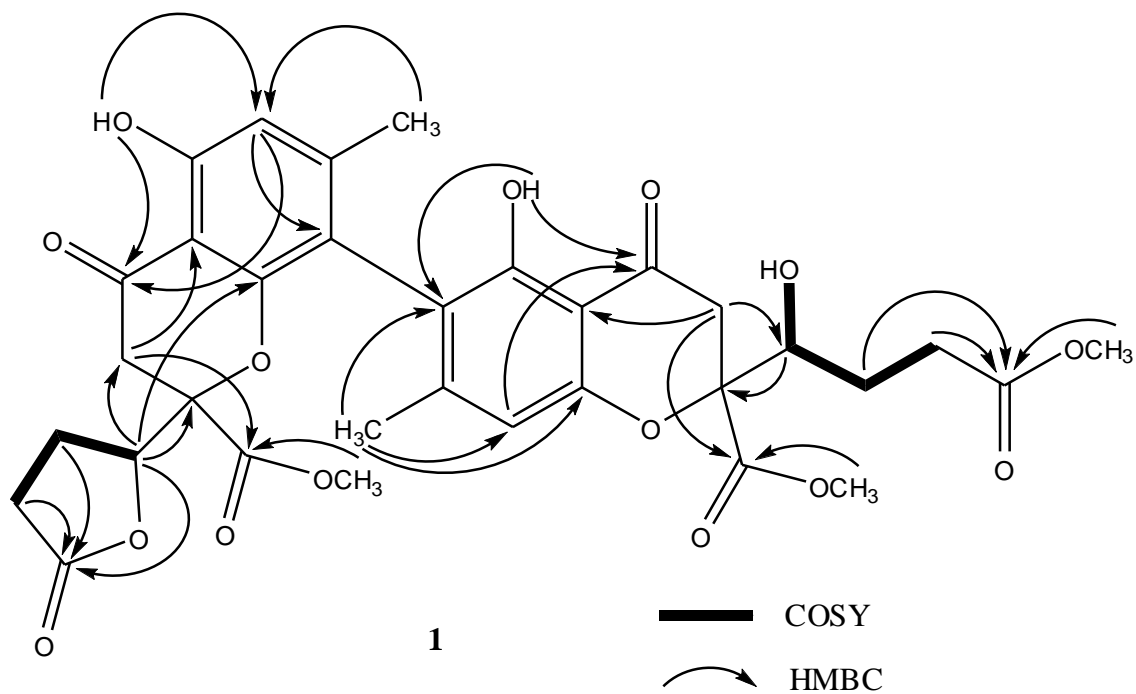
chromatograms revealed the pure MPA esters (2 mg = 52.8 % of *R*-MPA ester and 1.6 mg = 54.6 % of *S*-MPA ester, respectively).

Molecular Modeling. All models were calculated employing conformation search (Boltzman jump) and a standard force field as implemented in the Cerius2 4.0 (MSI) molecular modeling software package. Models were further refined with 1500 iterations of minimization. Calculations were performed using a Silicon Graphics O2 workstation (Irix 6.5.6).

Determination of Potential Cancer Chemopreventive Activities. Homogenates of H4IIE rat hepatoma cells induced for 39 h with the CYP1A inducer β -naphthoflavone (β -NF) at a concentration of 10 μ M were used as an enzyme source to measure CYP1A activity. The rate of time-dependent dealkylation of 3-cyano-7-ethoxycoumarin (CEC) to 3-cyano-7-hydroxycoumarin (CHC) was determined fluorimetrically in 96-well plates for 45 min at 37°C using a SpectraMax Gemini XS fluorescence reader (Molecular Devices, excitation 409 nm, emission 451 nm, cutoff 435 nm).^[8] Activity of solvent control C: 36 ± 7 pmol min⁻¹ mg⁻¹ of protein ($n=3-4$). Means significantly different from control (* $P < 0.01$, ** $P < 0.001$) using ANOVA with Holm-Sidak test for multiple comparison with $n=3-4$. Inhibition constants were generated from Lineweaver-Burk-, Dixon- and Cornish-Bowden plots of the results of kinetic experiments with 2.5 μ M, 5 μ M and 10 μ M CEC, respectively, as a substrate. The IC₅₀ value of α -naphthoflavone (α -NF), a known CYP1A inhibitor employed as a positive control, was 0.005 ± 0.001 μ M ($n=4$).

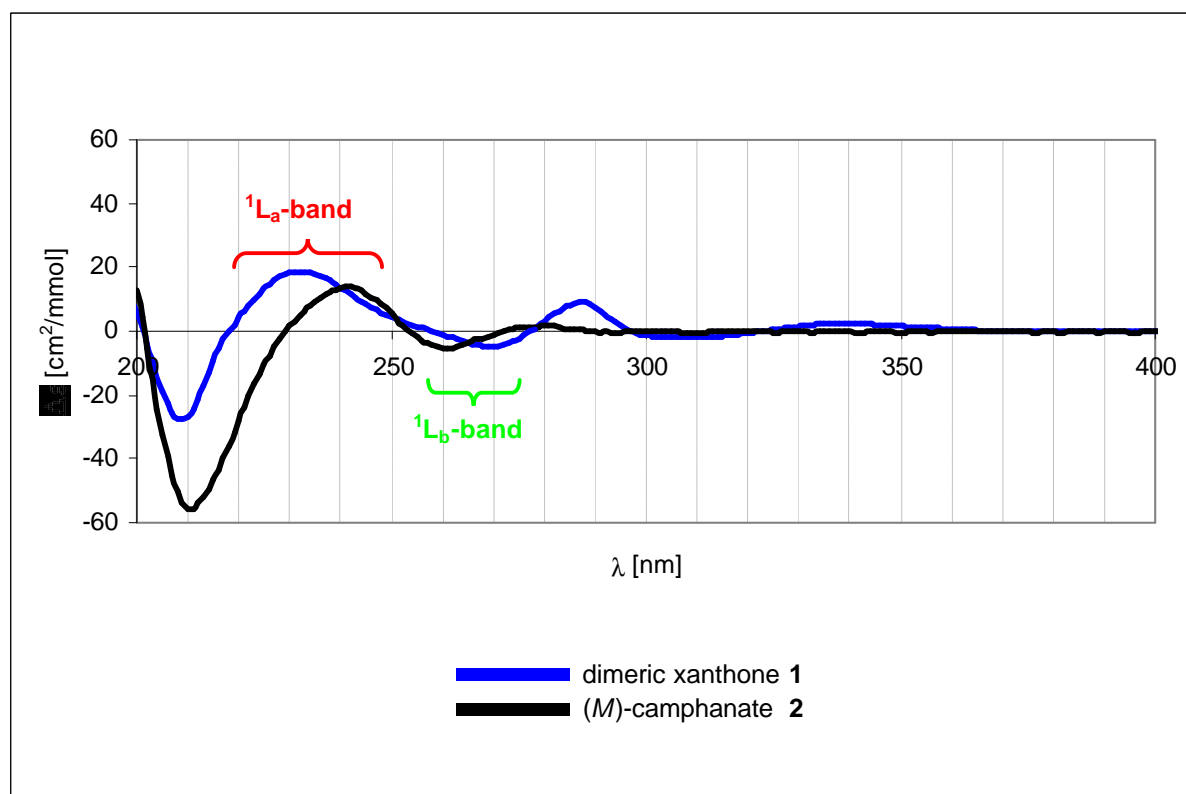
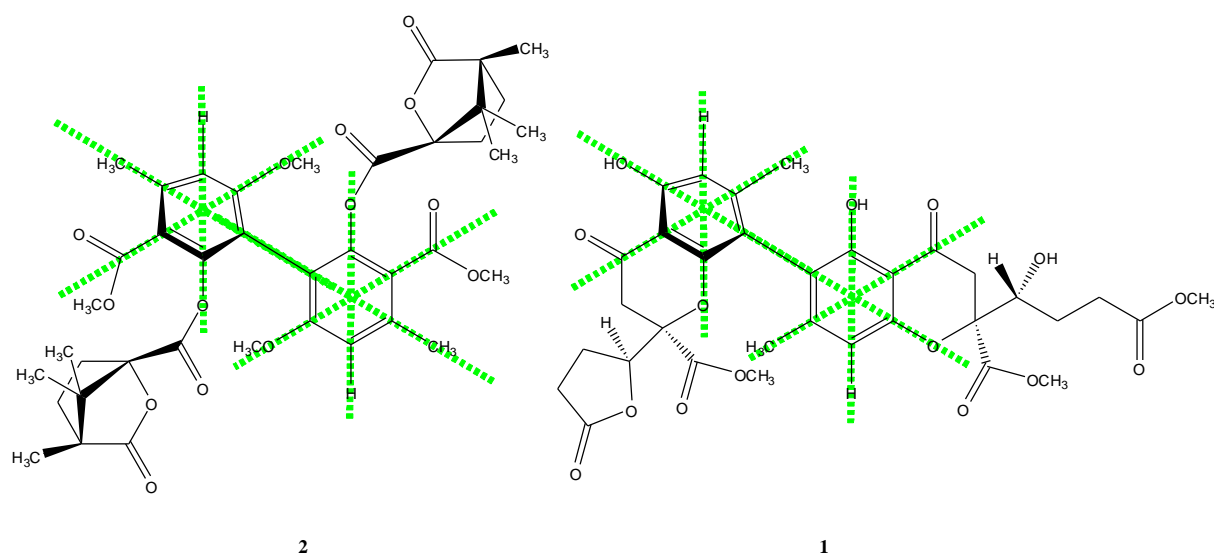
For the detection of phase 2 enzyme inducers, QR activity was measured in cultured Hepa 1c1c7 murine hepatoma cells (2×10^4 cells mL⁻¹) after a 48 h induction period by the NADPH-dependent menadiol-mediated reduction of MTT [3-(4,5-dimethylthiazo-2-yl)-2,5-diphenyltetrazolium bromide] to a blue formazan as described previously.^[8] Induction of QR activity was calculated from the ratio of specific enzyme activities of compound-treated cells in comparison with a solvent control, and CD values (concentration required to double the specific QR activity in μ M) were generated. Sulforaphane, an isothiocyanate from broccoli, was used as a positive control with a CD value of 0.19 ± 0.05 μ M ($n=3$). Specific activities of untreated controls were 26 ± 3 nmol min⁻¹ mg⁻¹ protein ($n=4$).

Figure S1. Important ^1H , ^1H -COSY and ^1H , ^{13}C long range (HMBC) correlations of compound **1**.



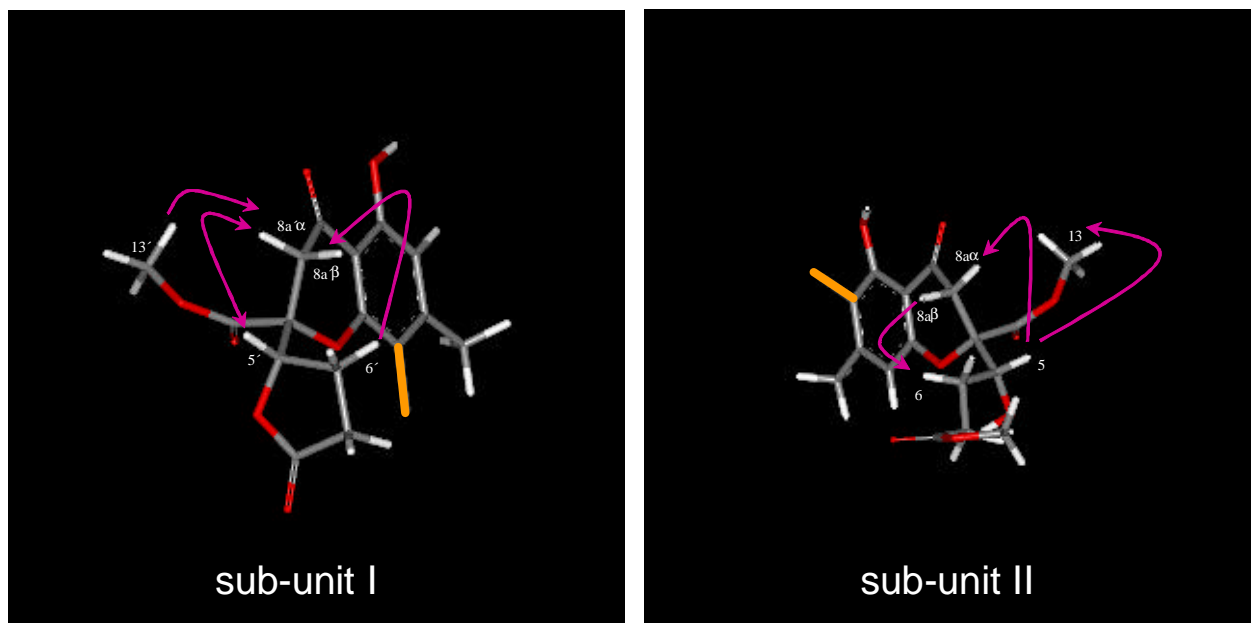
Supporting information

Figure S2. CD spectra of compound **1** and the reference (*M*)-orsellinic acid camphanate (**2**) in MeCN with sector rule for the 1L_b -Cotton effect (drawn through the ring carbons)^[10,11] for the benzene chromophores.



Supporting information

Figure S3. Selective gradient NOEs (purple arrows) depicted in autonomous 3D models (Cerius2) of both monomeric sub-units disconnected at the biphenyl axis (brownish bond).



Supporting information

Figure S4. Selective gradient NOEs (arrows) between the two sub-units of **1**.

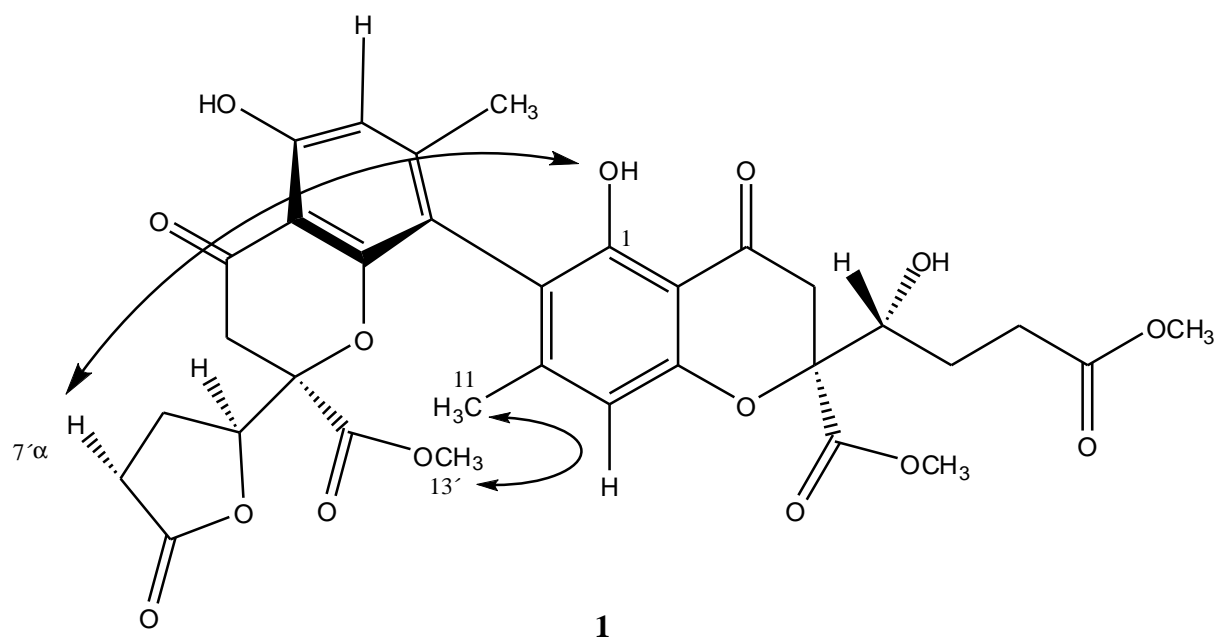
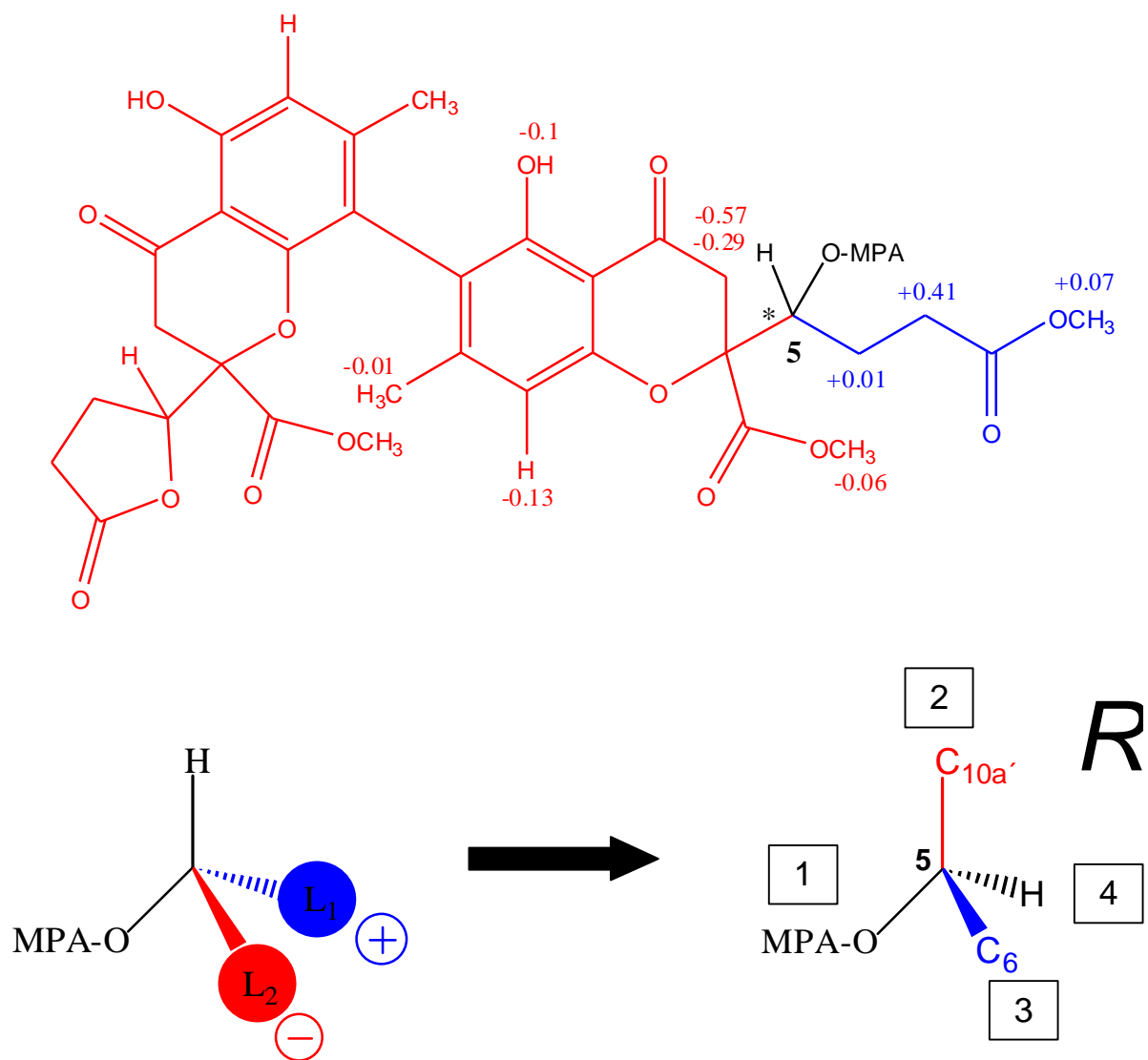
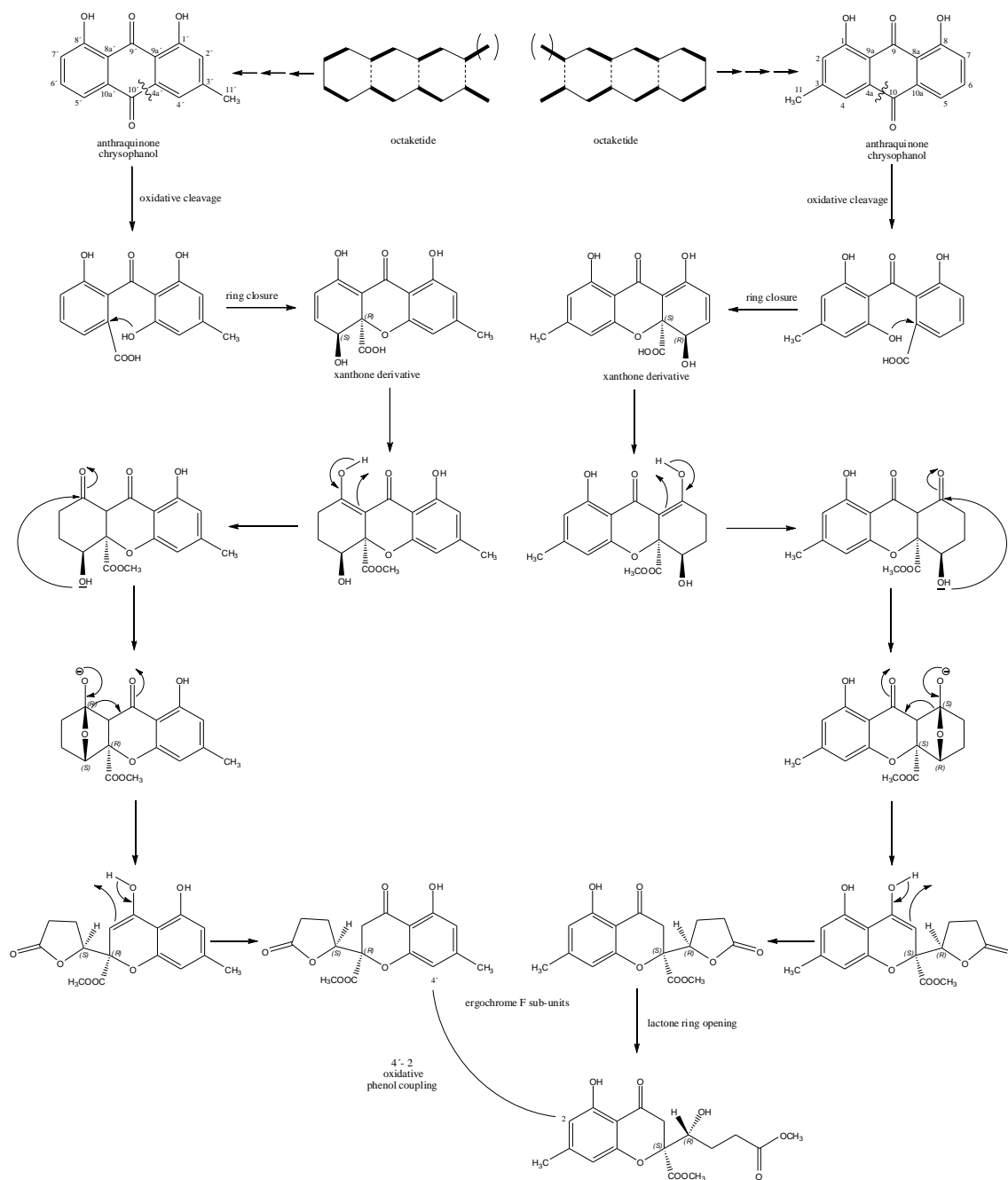


Figure S 5. Deduction of the absolute configuration at C-5 using modified Mosher's method with δd^{RS} -values of MPA esters of compound **1**.



Supporting information

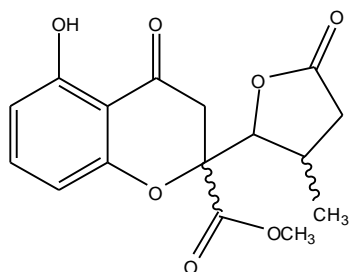
Figure S6. Proposed biosynthesis for compound **1**.



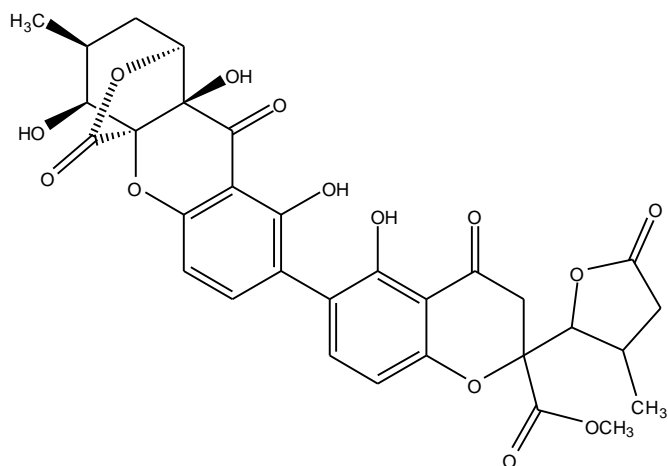
Supporting information

Figure S7. Structures of fungal metabolites related to **1**.

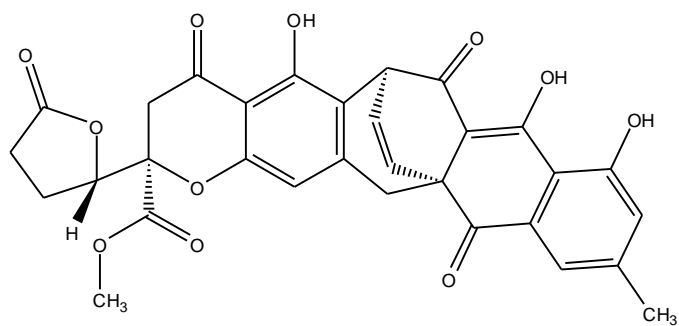
ergochrome F unit^[15]:



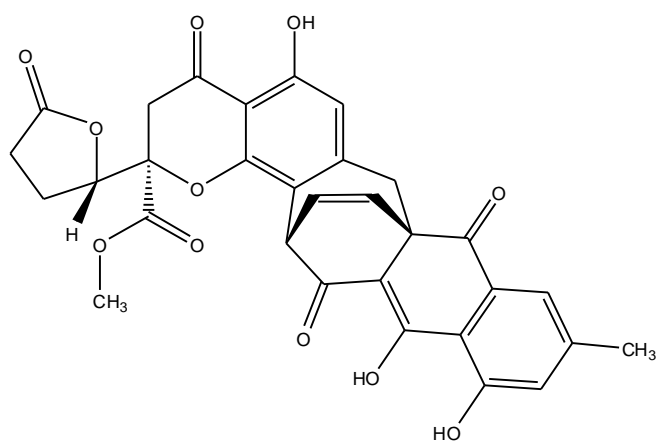
ergoxanthin^[15]:



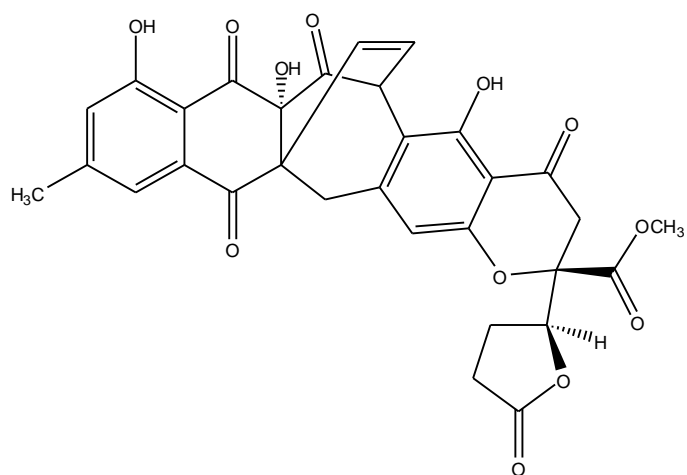
xanthoquinodin A3^[14]:



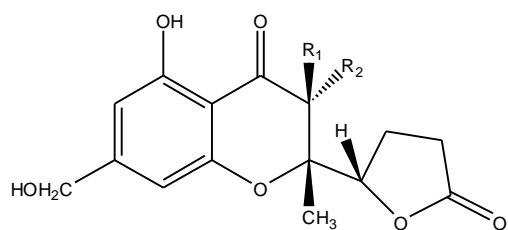
xanthoquinodin B3^[14]:



chaetomanone^[16]:



lachnone 3, 4 and 5^[17]:

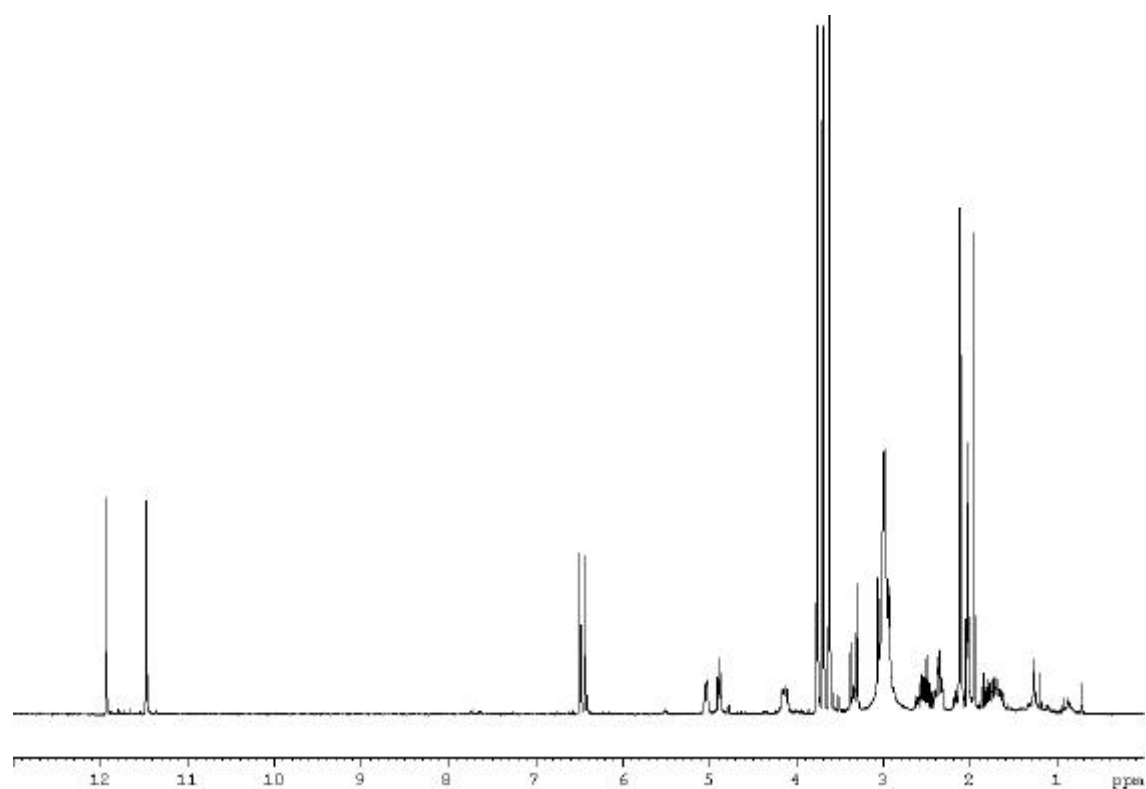


3: $R_1 = R_2 = H$

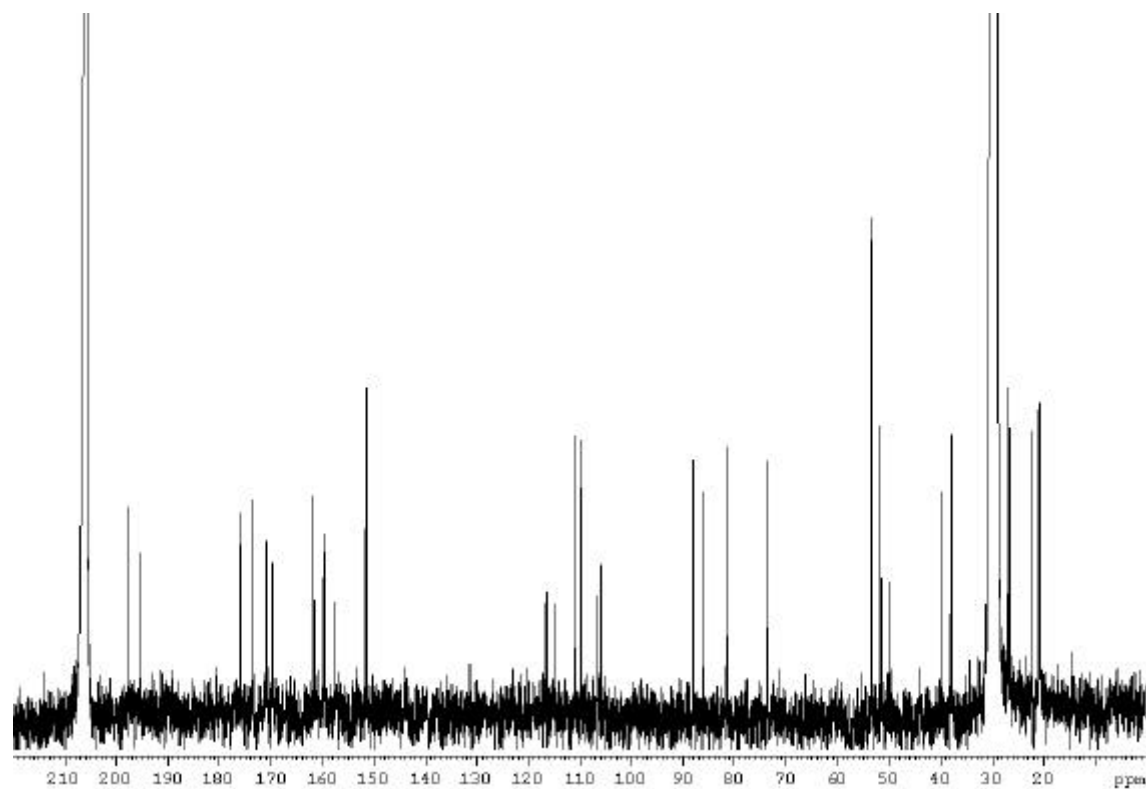
4: $R_1 = OH$; $R_2 = H$

5: $R_1 = -CH_2COCH_3$; $R_2 = OH$

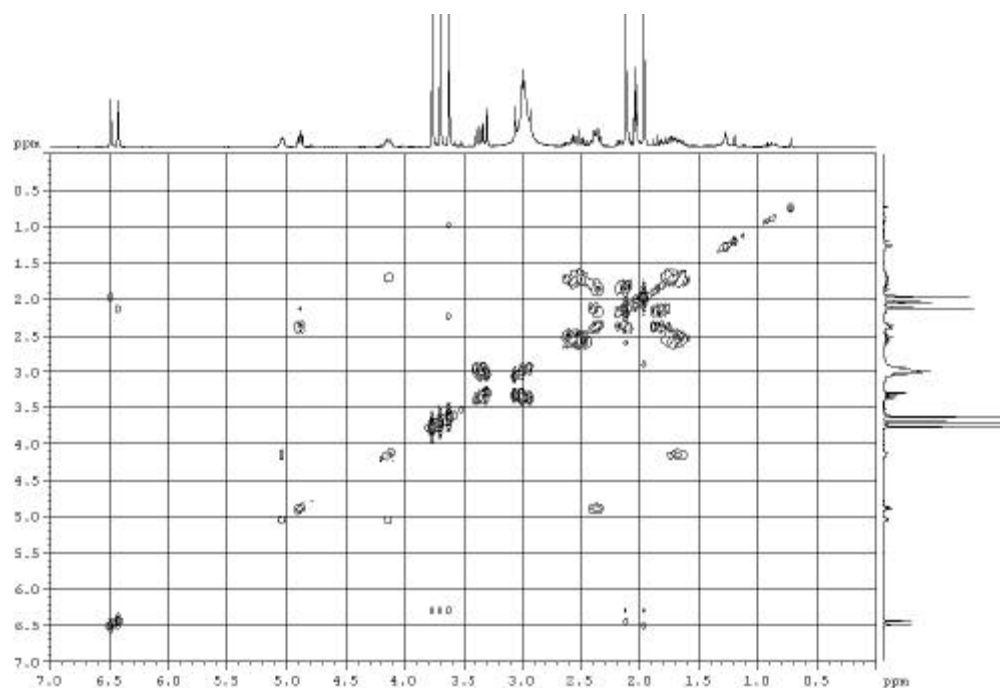
Supporting information – ^1H NMR spectrum (300 MHz in $[\text{D}_6]\text{acetone}$) of the new compound **1**.



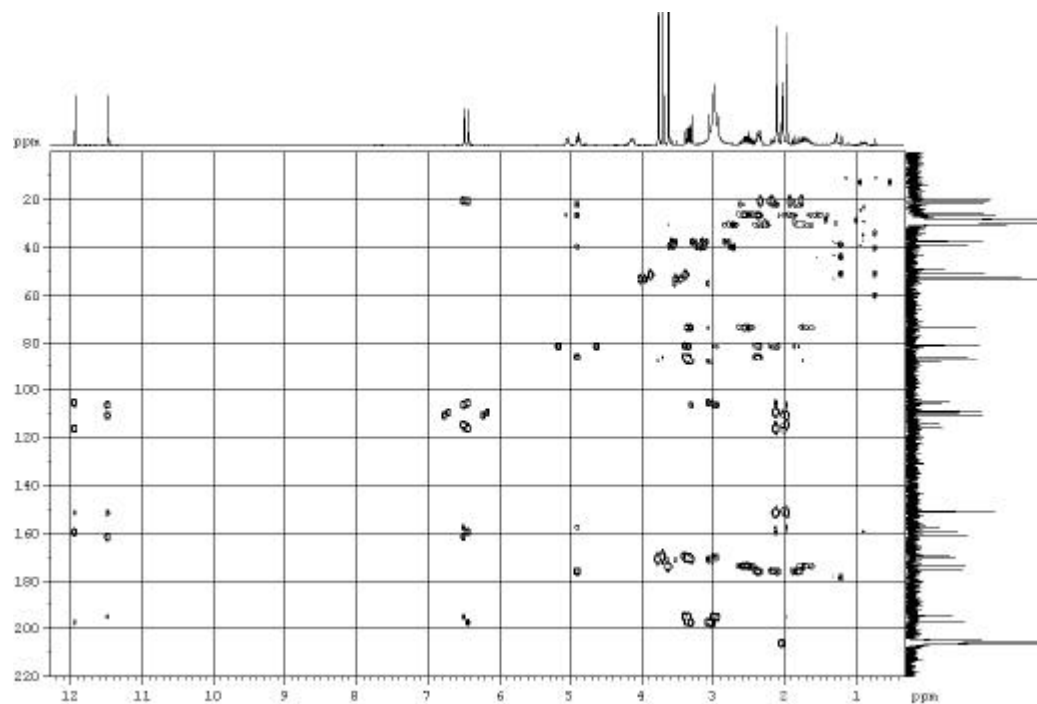
Supporting information – ^{13}C NMR spectrum (75 MHz in $[\text{D}_6]\text{acetone}$) of the new compound **1**.



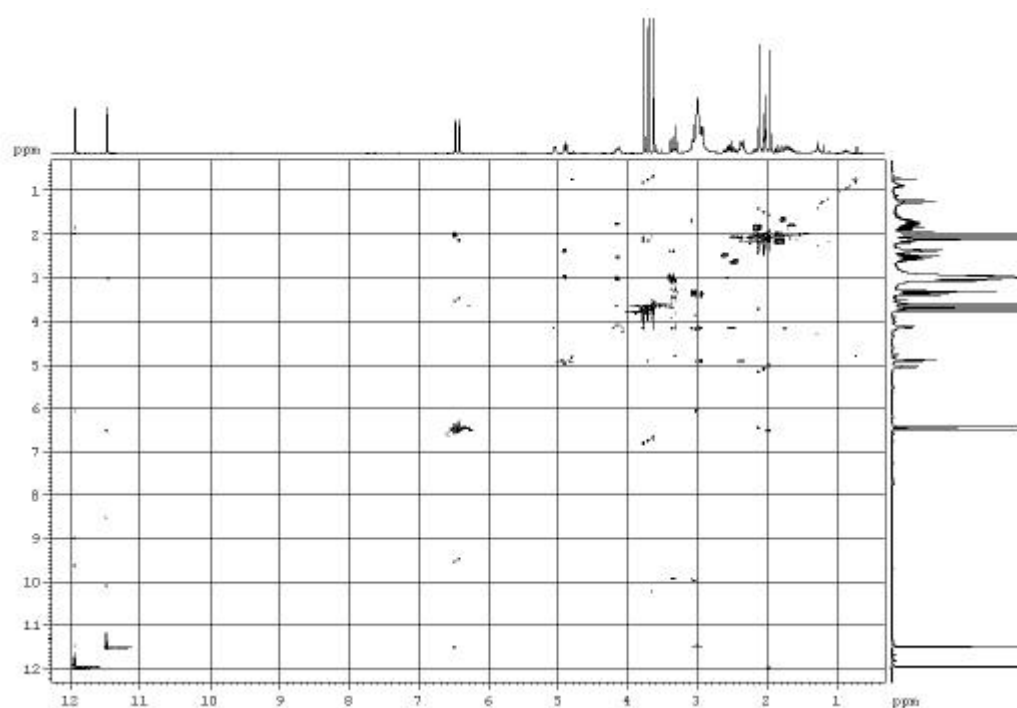
Supporting information $^{-1}\text{H}, ^1\text{H}$ -COSY spectrum (300 MHz in $[\text{D}_6]\text{acetone}$) of the new compound **1**.



Supporting information $^{-1}\text{H}, ^{13}\text{C}$ -HMBC spectrum (300 MHz in $[\text{D}_6]\text{acetone}$) of the new compound **1**.



Supporting information –¹H-¹H-NOESY spectrum (300 MHz in [D₆]acetone) of the new compound **1**.



Supporting information

Table S2. ¹H NMR spectral data for sub-unit II of (*R*) and (*S*)-MPA products of **1** with calculated δd^{RS} values.

no. ^a	(<i>R</i>)-MPA product <i>d</i> ¹ H ppm (mult.) ^b	(<i>S</i>)-MPA product <i>d</i> ¹ H ppm (mult.) ^b	δd^{RS}
4	6.27 (s)	6.40 (s)	-0.13
5	5.61 (d)	5.59 (d)	+0.02
6	1.77 (m)	1.76 (m)	+0.01
7	2.40 (m)	1.99 (m)	+0.41
8aa	2.75 (d)	3.04 (s)	-0.29
8aβ	2.47 (d)	3.04 (s)	-0.57
11	2.11 (s)	2.12 (s)	-0.01
13	3.74 (s)	3.80 (s)	-0.06
14	3.70 (s)	3.63 (s)	+0.07
OH-1	11.66 (s)	11.76 (s)	-0.10

^a Position of proton atom. ^b CDCl₃, 300/75.5 MHz.